

with relative atomic masses 1993 according to IUPAC

1 (I)	2 (II)	3	4	5	6	7	8	9	10	11	12	13 (III)	14 (IV)	15 (V)	16 (VI)	17 (VII)	18 (VIII)
1 H 1.00794																	2 He 4.002602
3 Li 6.941	4 Be 9.012182	<div>Atomic number Symbol Relative atomic mass*</div>										5 B 10.811	6 C 12.011	7 N 14.00674	8 O 15.9994	9 F 18.9984032	10 Ne 20.1797
11 Na 22.989768	12 Mg 24.3050											13 Al 26.981539	14 Si 28.0855	15 P 30.973762	16 S 32.066	17 Cl 35.4527	18 Ar 39.948
19 K 39.0983	20 Ca 40.078	21 Sc 44.955910	22 Ti 47.867	23 V 50.9415	24 Cr 51.9961	25 Mn 54.93805	26 Fe 55.845	27 Co 58.93320	28 Ni 58.6934	29 Cu 63.546	30 Zn 65.39	31 Ga 69.723	32 Ge 72.61	33 As 74.92159	34 Se 78.96	35 Br 79.904	36 Kr 83.80
37 Rb 85.4678	38 Sr 87.62	39 Y 88.90585	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.94	43 Tc (98)	44 Ru 101.07	45 Rh 102.90550	46 Pd 106.42	47 Ag 107.8682	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.760	52 Te 127.60	53 I 126.90447	54 Xe 131.29
55 Cs 132.90543	56 Ba 137.327	La– Lu	72 Hf 178.49	73 Ta 180.9479	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.217	78 Pt 195.08	79 Au 196.96654	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.98037	84 Po (209)	85 At (210)	86 Rn (222)
87 Fr (223)	88 Ra (226)	Ac– Lr	104 Rf (261)	105 Db (262)	106 Sg (263)	107 Bh (262)	108 Hs (265)	109 Mt (266)	**								

57 La 138.9055	58 Ce 140.115	59 Pr 140.90765	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.965	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
89 Ac (227)	90 Th (232.0381)	91 Pa (231.03588)	92 U (238.0289)	93 Np (237)	94 Pu (239)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

Relative atomic mass based on  $A_r(^{12}\text{C}) \equiv 12$  (after IUPAC “Atomic Weights of the Elements 1993”, *Pure and Applied Chemistry*, **1994**, 66(12), 2423–2444). For elements which lack stable isotope(s) is the mass number for the most stable isotope given in parentheses, or for Th, Pa and U the relative atomic mass given by IUPAC for the isotopic mixture present on Earth.

\*\* Chemical symbols for elements 104 – 109 according to IUPAC “Names and Symbols of Transfermium Elements (IUPAC Recommendations 1997)”, *Pure and Applied Chemistry*, **1997**, 69(12), 2471-2473.